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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO
09/656,777	09/07/2000	Junji Kuyama	09793822-0409	1570
26263	7590 10/05/2004		EXAMINER	
SONNENSCHEIN NATH & ROSENTHAL LLP			WILLS, MONIQUE M	
	P.O. BOX 061080 WACKER DRIVE STATION, SEARS TOWER		ART UNIT	PAPER NUMBER
	IL 60606-1080		1746	·
			DATE MAILED: 10/05/2004	4

Please find below and/or attached an Office communication concerning this application or proceeding.

			4				
	Application No.	Applicant(s)	7				
	09/656,777	KUYAMA ET AL.					
Office Action Summary	Examiner	Art Unit					
	Monique M Wills	1746					
The MAILING DATE of this communication ap Period for Reply	pears on the cover sheet	with the correspondence address					
A SHORTENED STATUTORY PERIOD FOR REPL THE MAILING DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1.1 after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a rep - If NO period for reply is specified above, the maximum statutory period - Failure to reply within the set or extended period for reply will, by statute Any reply received by the Office later than three months after the mailin earned patent term adjustment. See 37 CFR 1.704(b).	136(a). In no event, however, may ly within the statutory minimum of t will apply and will expire SIX (6) Me, cause the application to become	a reply be timely filed nirty (30) days will be considered timely. DNTHS from the mailing date of this communication ABANDONED (35 U.S.C. § 133).					
Status							
1) Responsive to communication(s) filed on 29 S	September 2004.						
2a)⊠ This action is FINAL . 2b)☐ This	s action is non-final.						
	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.						
Disposition of Claims							
4) ☐ Claim(s) 1-9 and 23-34 is/are pending in the a 4a) Of the above claim(s) 1-9 is/are withdrawn 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 23-34 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/o	from consideration.						
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 9) The specification is objected to by the Examine 10) The drawing(s) filed on <u>07 September 2000</u> is the Applicant may not request that any objection to the Replacement drawing sheet(s) including the correct 11) The oath or declaration is objected to by the Example 11. 	are: a)⊠ accepted or b) drawing(s) be held in abey tion is required if the drawir	ance. See 37 CFR 1.85(a). g(s) is objected to. See 37 CFR 1.121(d)).				
Priority under 35 U.S.C. § 119							
a) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority document 2. Certified copies of the priority document 3. Copies of the certified copies of the priority document application from the International Bureau * See the attached detailed Office action for a list	ts have been received. ts have been received in rity documents have bee u (PCT Rule 17.2(a)).	Application No n received in this National Stage					
Attachment(s)) Notice of References Cited (PTO-892)	4) ☐ Interview	Summary (PTO-413)					
Notice of Draftsperson's Patent Drawing Review (PTO-948) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Paper No(s)/Mail Date	Paper No	(s)/Mail Date Informal Patent Application (PTO-152)					

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DETAILED ACTION

Request for Continued Examination

A Request for Continued Examination (RCE) under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on September 30, 2004 has been entered.

The following actions are overcome:

- The objection of claim 27, because of the misspelling of polyvinylidene fluoride
- The rejection of claims 23 & 24, under 35 U.S.C. § 112, second paragraph as being indefinite.

The following rejections are maintained:

- Claims 23-25, 29-31 & 33-34 under 35 U.S.C. § 102(e) as being anticipated by Isoyama et al., U.S. Patent 6,093,503.

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Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(e) the invention was described in (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.

Claims 23-25, 29-31 & 33-34 are rejected under 35 U.S.C. 102(e) as being anticipated by Isoyama et al., U.S. Patent 6,093,503.

With respect to claims 23 & 29, Isoyama teaches a method of making a positive electrode active material comprising: mixing a first ingredient of Ketjen Black and 90% by weight of lithium manganese oxide (Example 22); press molding the mixture (col. 12, lines 5-10); sintering the mixture in a temperature range from 300 to 1200°C embracing Applicant's range not lower than 600°C and not higher than 850°C (col. 7, lines 57-68); wherein the positive electrode is a lithium composite manganese oxide comprising an aggregate (col.2, lines 12-20) of primary particles having a grain diameter of 1 to 20 microns and the negative electrode is a metallic lithium (col. 2, lines 12-20). Further concerning claim 23, the lithium composite oxide is LiMn₂O₄ meeting the general formula Li_xMn_{2-y}M_yO₄ where x=1 and y=0. Further concerning claim 29, the negative electrode includes coke and an organic polymer compound sintered body (col. 23, lines 8-20). The limitation in claims 23 & 29, with respect to the specific surface area measured by BET between 0.2m²/g and 2m²/g, is considered to be an inherent property of the cathode material as set forth in the prior art, because Isoyama employees the same lithium manganese oxide material with the same primary particle size as set forth by Applicant. The limitation in claims 23, 29 & 30, with respect to the negative

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electrode material reversibly doping and dedoping lithium is considered to be an inherent property of the negative electrode as set forth in the prior art, because Isoyama employs the same lithium anodic material set forth by Applicant. Additionally, "products of identical chemical composition can not have mutually exclusive properties." A chemical composition and its properties are inseparable. Therefore, if the prior art teaches the identical chemical structure, the properties applicant discloses and/or claims are necessarily present. In re Spada, 911 F.2d 705, 709, 15 USPQ 2d 1655, 1658.

With respect to claims 24 & 25, the spinel LiMn₂O₄ (col. 6, lines 25-30) has a primary particle size of 1 to 20 microns, embracing a primary particle diameter of 0.5 to 3 microns. Specific particle sizes of about 1 to 3 microns are exemplified in column 29, lines 24-50. With respect to claim 31, the negative electrode is metallic lithium (col. 2, lines 30-40). With respect to claim 33, the electrolyte salts include LiClO₄, LiBF₆, LiPF₆, LiCF₃SO₃ and LiAsF₆ (col. 5, lines 40-45). Regarding claim 34, the electrolyte is dissolved in an organic solvent selected from propylene carbonate, diethyl carbonate and gamma-butyrolactone (col. 5, lines 41-46).

Therefore, the limitations are anticipated by the prior art set forth.

Claim Rejections ~ 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

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Claims 26-28 & 32 are rejected under 35 U.S.C. 103(a) as being unpatentable over Isoyama et al., U.S. Patent 6,093,503 in view of Miyasaka U.S. Patent 5,869,208.

Isoyama teaches a method of making a positive active material as described in the 35 U.S.C. § 102(e) rejection hereinabove. The method includes creating a slurry by kneading an admixture of graphite and polyvinylidene fluoride (col. 5, lines 35 & col. 39, lines 10-20) with LiMnO₂ dissolved in a liquid phase (col. 39, lines 5-20). The lithium oxide, conductive agent and binder are mixed in a weight ratio of 9: 0.6 to 0.4 (col. 39, lines 10-20). With respect to claim 28, cathode material is applied to an aluminum foil current collector (col. 39, lines 10-15) with a thickness of 0.02 to 200 microns.

Isoyama is silent to created a slurry of active material, binder and conductive again (claims 27 & 32), employing 86% lithium composite manganese oxide (claim 276) and 10% graphite (claims 27 & 32). The reference is also silent to pulverizing the sintered mixture (claim 26).

Miyasaka teaches that it is conventional to create a slurry of electrode material prior to coating on a current collector (col. 123, lines 5-15). The electrode material includes lithium manganese oxide, a binder and conductive agent (col. 12, lines 5-`5). The reference also teaches pulverizing to increase the specific surface area of the active material (col. 11, lines 20-30).

It would have been obvious to one having ordinary skill in the art at the time the instant invention was made to employ the slurry preparation of Miyasaka in the method of Isoyama, in order to facilitate coating electrode material on the current collector. The skilled artisan recognizes that a slurry would increase malleability of the active material thereby improving coating ability of said material on the current collector (claims 27 & 32).

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With respect to pulverizing the sintered electrode material (claim 26), the invention as a whole would have been obvious to one having ordinary skill in the art at the time the invention was made, because even though Isoyama is silent to pulverizing the active material, Miyasaka teaches that pulverization increases the specific surface area of the active material (col. 11, lines 20-30).

With respect to the amount of lithium manganese oxide, it would have been obvious to one of ordinary skill in the art at the time the time the invention was made to employ 86% by weight lithium manganese oxide since it has been held that discovering optimum value of a result effective variable involves only routine skill in the art. In re Boesch, 617 F 2d 2727, 205 USPQ 215 (CCPA 1980). The skilled artisan recognizes that the amount of active material directly effects the amount of voltage and current produced by the cell.

With respect to the amount of graphite, it would have been obvious to one of ordinary skill in the art at the time the invention was made to employ 10% by weight of graphite since it has been held that discovering optimum value of a result effective variable involves only routine skill in the art. In re Boesch, 617 F. 2d 272, 205 USPQ 215 (CCPA 1980). The skilled artisan recognizes that the amount of conductive agent directly effects conductivity of the electrode.

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Response to Arguments

Applicant asserts that Isoyama is not anticipatory because the reference does not teach a lithium composite manganese oxide with a particle diameter of *not less than 0.05 μm* and not greater than 10 μm, since Isoyama's range of 1 to 20 μm clearly suggests a primary particle diameter greater than 10 μm. Additionally, Isoyama teaches centering the mixture at a temperature in the range of 300 to 1200°C. However, such a range does not teach or suggest centering a mixture at a temperature *not lower than 600°C* and *not higher than 850°C*, since the range of 300°C to 1200°C clearly suggest a centering temperature that is lower than 600°C and higher than 850°C. This argument is not persuasive. According to M.P.E.P. § 2131.03, a specific example in the prior art within a claimed range, anticipates the range. Therefore, Isoyama anticipates the instant ranges, because the reference exemplifies a LiMn₂O₄ cathode material having a particle size of 1 μm being centered in air at 600°C for 24 hrs. See Example 4. Accordingly, the claimed in invention is properly anticipated by Isoyama with respect to claims 23-25, 29-31 & 33-34 and Isoyama in view of Miyasaka are obvious over claims 27, 28 & 32.

Conclusion

Any inquiry concerning this communication or earlier communications from the Examiner should be directed to Monique Wills whose telephone number is (571) 272-1309. The Examiner can normally be reached on Monday-Friday from 8:30am to 5:00 pm.

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If attempts to reach Examiner by telephone are unsuccessful, the Examiner's supervisor, Michael Barr, may be reached at 571-272-1414. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR.

Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov.Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

MICHAEL BARR SUPERVISORY PATENT EXAMI

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09/30/04